Preparation and characterization of stable nano-crystalline p-type Cu_2O semiconductors modified with different metal doping for their applications in photoelectrochemical water splitting for Hydrogen generation.

Funding Agency	BRNS-DAE
Sanctioned Amount	Rs. 24.51 Lakhs-
Project Duration	3 Years
Project Status	Continuing since March 2014

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Collaborative Institute(s) (if any): BARC, Mumbai

Brief Description of the Project

- 1. Synthesis of nanocrystalline p-type Cu₂O semiconductors and their characterization for photoelectrochemical water splitting applications.
- 2. Modifications of the materials with various metal ions to enhance their activity.
- 3. Further modification using hydrogen evolution co-catalyst.
- 4. Physical characterizations of the semiconductor.
- 5. Electrochemical characterizations of the semiconductor through linear sweep voltammetry, chronoampeometry and electrochemical impedance spectroscopy.

Keywords: Doped p-type Cu2O nano crystals, Photoelectrochemical water splitting application. Scanning electrochemical microscopy (SECM), Controlled Intensity Modulated Photo-Spectroscopy (CIMPS), Absorbed photon to current conversion efficiency (APCE), Incident photon to current conversion efficiency (IPCE)

Methodologies/Approaches Adopted (Simple description):

- 1. Synthesis of Cu₂O semiconductor through electrodeposition techniques.
- 2. Preparation and Characterization of Cu₂O semiconductors nanoparticles through hydrothermal techniques.
- 3. Development of suitable metal doped Cu₂O semiconductor thin film through chemical bath deposition techniques.

Project Highlights

- 1. Cu₂O is the potential photoactive material should be abundant and cheap to make PEC technology competitive with chemical production of hydrogen from coal or natural gas.
- 2. Synthesis of nano crystalline p-type Cu₂O is a cost effective and easy technique and it is a non toxic material.
- 3. Identification of suitable dopants for improved performance of p-type Cu₂O for photoelectrochemical water splitting application.
- 4. Modification of semiconductor surface by applying coating of other compounds as well as cocatalysts for better stability and enhanced activity.
- 5. The optimized material will generate hydrogen photoelectrochemically directly from water.

Project Achievements

Substrate Variation

We developed of p-Cu₂O thin films through cathodic electrodeposition technique at constant current of 0.1mA/cm^2 on Cu, Al and ITO substrates from basic CuSO₄ solution containing Triton X-100 as the surfactant at 30-35°C. The photoelectrochemical reduction of water (H₂O \rightarrow H₂) in pH 4.9 aqueous solutions over the different substrates are found to be of the order of Cu > Al > ITO and the highest current of 4.6mA/cm^2 has been recorded over the Cu substrate even at a low illumination of 35mW/cm^2 , which is significantly higher than the value (2.4mA/cm^2 on Au coated FTO or 4.06mA/cm^2 on Cu foil substrate at an illumination of 100mW/cm^2) reported in literature.



Fig. 2: Characterization of Cu₂O Thin Film

Modification of Cu2O

For the first time we found out that, photoelectrochemical performance of the cathodically electrodeposited Cu₂O thin film improves dramatically by ~2 fold when BiNPs are added in the form of suspension or acts as a matrix coated over ITO glass. Further addition of an optimized amount (10 nM) of Bi3+ also facilitate the hydrogen evolution reaction of Cu₂O. Maximum photocurrent for Cu2O film developed form three different condition are as: (i) -5.2 mA/cm² for -4.9 ITO/BiNPfilm/Cu2O, (ii) mA/cm² for ITO/BiNPsus/Cu2O, (iii) -3.7 mA/cm2 for ITO/Bi3+ion-Cu2O whereas that for pure Cu₂O on ITO appears as -2.6 mA/cm2. This is the highest reported photocurrent of Cu2O on any conducting glass substrates without hydrogen evolution catalyst.



Thin Film











Publications (Please follow any one specific bibliographic style)

- Sanjib Shyamal, Paramita Hajra, Harahari Mandal, Jitendra Kumar Singh, Ashis Kumar Satpati, Surojit Pande, and Chinmoy Bhattacharya, Effect of Substrates on the Photoelectrochemical Reduction of Water over Cathodically Electrodeposited p-Type Cu2O Thin Films, ACS Appl. Mater. Interfaces 2015, 7, 18344–18352.
- 2. Technical paper entitled "Catalytic role of Bi on electrodeposited p-type Cu2O thin films for photoelectrochemical production of H2 from water" presented at "Current Trends in Analytical Chemistry" organized by BARC-AEACI, Mumbai on May 26-29, 2015.
- Technical paper entitled "Galvanostatic Electrodeposition of p-type Cu2O thin Films: Effect of Cu / Al / Conducting Glass Substrates for Photoelectrochemical Hydrogen Production" " presented at "Young Scientist's colloquium" organized by MRSI, Kolkata chapter, Kolkata on September 11, 2015.
- Technical paper entitled "Modification of Cu2O semiconductor using Bismuth (Nanoparticles or Ions) for Photoelectrochemical Hydrogen Production" presented at "International Conference on Materials for the Millennium, MATCON-2016" organized by Department of Applied Chemistry, CUSAT, Kochi on January 14-16, 2016, ISBN 978-93-80095-738.
- Sanjib Shyamal, Paramita Hajra, Harahari Mandal, Aparajita Bera, Debasis Sariket, Ashis Kumar Satpati, Sukumar Kundu and Chinmoy Bhattacharya, Benign role of Bi on an electrodeposited Cu2O semiconductor towards photo-assisted H₂ generation from water, J. Mater. Chem. A, 2016, 4, 9244-9252.

Facilities Developed

- 1. Photoelectrochemical instrument set up with Controlled Intensity Modulated Photo-Spectroscopy (CIMPS) consisting of software & hardware of electrometer and computer controlled high power Xenon arc lamp source and monochromatic beam intensity.
- 2. Computer controlled high power Xenon arc lamp source (white light) with monochromatic beam intensity for dynamic measurements on photo-active systems.

Project Staff

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Plan of Future Project Proposal based on the Current Project:

- 1. Production of H₂ and O₂ through visible-light driven photocatalytic water splitting process using dual semiconductors-graphene mediator composite materials in Z-scheme device.
- 2. Improved design for the Z-scheme device for production of H₂ from water splitting process, as measured through Membrane Electrode Assemble (MEA) using in the Fuel cell Technology.